

Preliminary communication

Stereoselective synthesis of methyl 4-*C*-acetyl-6-deoxy-2,3-*O*-methylene-D-galactonate and -D-gluconate. Determination of the D-*galacto* configuration of methyl eurekanate by synthesis

JUJI YOSHIMURA and MASAFUMI MATSUZAWA

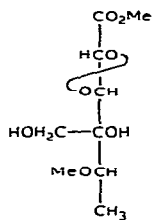
Laboratory of Chemistry for Natural Products, Faculty of Science, Tokyo Institute of Technology, Nagatsuta, Midoriku, Yokohama 227 (Japan)

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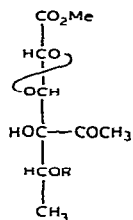
Oligosaccharide antibiotics, everninomicins¹ and flambamycin² of the orthosomycin family contain a characteristic 4-*C*-substituted 2,3-*O*-methylene-aldonolactone which is attached at a terminal position by an acetal interlinkage. The configuration of the lactone in everninomicins B and D was determined by X-ray analysis³ and characterized as the corresponding methyl aldionate⁴. Recently, we have synthesized the methyl ester, namely, methyl 6-deoxy-4-*C*-(hydroxymethyl)-5-*O*-methyl-2,3-*O*-methylene-L-idonate (**1**) from L-arabinose⁵. As the configuration of methyl eurekanate (methyl 4-*C*-acetyl-6-deoxy-2,3-*O*-methylenhexonate) from flambamycin was ambiguous, the D-*galacto* (**2**) and D-*gluco* (**4**) diastereomers were stereoselectively synthesized from D-glucose, on the assumption that the configurations of the carbon atoms bearing the characteristic 2,3-methylenedioxy group in **1** and methyl eurekanate are the same. As a result, the configuration of methyl eurekanate has now been determined to be D-*galacto*.

Treatment with sodium hydride of benzyl 4,6-*O*-benzylidene- α -D-glucopyranoside in *N,N*-dimethylformamide and dichloromethane⁶ and separation of the products on a column of silica gel using 8:1 hexane–ethyl acetate, gave the corresponding 2,3-*O*-methylene derivative (**5**), m.p. 107–108°, $[\alpha]_D^{+123}$ (*c* 1.0, CHCl₃), and two dimers, **6**, m.p. 111–113°, $[\alpha]_D^{+171}$ (*c* 0.98, CHCl₃), and **7**, m.p. 114–116°, $[\alpha]_D^{+196}$ (*c* 0.8, CHCl₃) in 43, 6.2, and 7.2% yield, respectively. The configurations of **6** and **7** were deduced from the methylene proton signals in their ¹H-n.m.r. spectra (**6**: δ 4.75 and 4.96, ABq, *J* 7.0 Hz; **7**: δ 4.70 and 4.90, each s). Partial hydrolysis of **5** with 70% acetic acid for one day at room temperature gave the *O*-debenzylidenated product **8**, m.p. 108–110°, $[\alpha]_D^{+176}$ (*c* 1.0, CHCl₃) in 64% yield. Monotosylation of **8** in pyridine with *p*-toluenesulfonyl chloride gave the 6-*O*-tosyl derivative (**9**), m.p. 80–82°, $[\alpha]_D^{+78.9}$ (*c* 1.2, CHCl₃), in 81% yield. Reduction of **9** in dimethyl sulfoxide with sodium borohydride gave the 6-deoxy derivative (**10**) as a syrup, $[\alpha]_D^{+154}$ (*c* 0.85, CHCl₃), in 81% yield. Oxidation of **10** with dimethyl sulfoxide–trifluoroacetic anhydride⁷ gave a quantitative yield of benzyl 6-deoxy-2,3-*O*-methylene- α -D-xylo-hexopyranosid-4-ulose (**11**) as a syrup: ¹H-n.m.r. data (CDCl₃): δ ~7.38 (m, 5 H, Ph), 5.47 (d, 1 H, *J*_{1,2} 3.0 Hz, H-1), 5.19 and 5.10 (each d, 2 H, *J*_{gem} 0.8 Hz, OCH₂O), 4.82

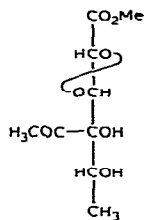
(s, 2 H, CH₂Ph), 4.70 (dd, 1 H, $J_{3,5}$ 1.0 Hz, H-3), 4.12 (dq, 1 H, $J_{5,6}$ 3.5 Hz, H-5), 3.63 (dd, 1 H, $J_{2,3}$ 10.5 Hz, H-2), and 1.35 (d, 3 H, H-6). Syrupy **11** was used for the next reaction without purification.



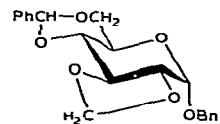
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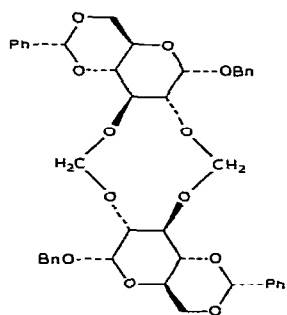
2 R = H
3 R = Ac



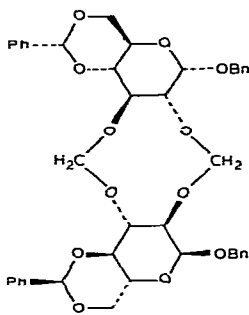
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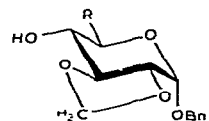
5 Ph = C₆H₅
Bn = C₆H₅CH₂



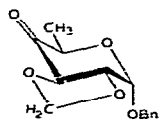
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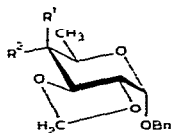
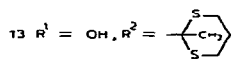
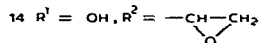
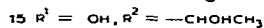
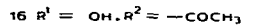
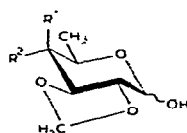
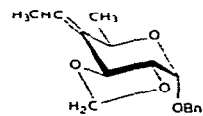
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8 R = CH₂OH
9 R = CH₂OTs
10 R = Me



11

12 R¹ = OH, R² = -CH=CH₂13 R¹ = OH, R² =14 R¹ = OH, R² = -CH-CH₂15 R¹ = OH, R² = -CHOHCH₃16 R¹ = OH, R² = -COCH₃17 R¹ = -CHOHCH₃, R² = OH18 R¹ = -COCH₃, R² = OH19 R¹ = OH, R² = -COCH₃20 R¹ = -COCH₃, R² = OH

21

Reaction of **11** with vinylmagnesium bromide in oxolane (tetrahydrofuran) at room temperature, and separation of the products on a column of silica gel with 4:1 hexane—ethyl acetate, gave benzyl 6-deoxy-2,3-*O*-methylene-4-*C*-vinyl- α -D-galactopyranoside (**12**) as a syrup, $[\alpha]_D +141^\circ$ (*c* 6.8, CHCl_3), and the 4-epimer of **12** as a syrup, $[\alpha]_D +167^\circ$ (*c* 2.3, CHCl_3) in the ratio of 15.9:1 in 55.4% yield. These configurations were deduced from the chemical shifts of the α -carbon atoms in the vinyl groups: ^{13}C -n.m.r. data, **12**, 137.6; 4-epimer, 132.7 p.p.m. (ref. 8). A similar reaction of **11** with 2-lithio-2-methyl-1,3-dithiane⁹ gave, exclusively, the corresponding D-*galacto* derivative (**13**) as a syrup in 40% yield, $[\alpha]_D +116^\circ$ (*c* 0.8, CHCl_3); ^{13}C -n.m.r. of C-2 of the 1,3-dithianyl group: 58.36 p.p.m. (ref. 10). Epoxidation of **12** with *m*-chloroperoxybenzoic acid in 1,2-dichloroethane at 80° gave a mixture of the corresponding (*S*)- and (*R*)-epoxides (**14**) in 46% yield; this was converted into the 4-*C*-acetyl derivative (**16**), a syrup, $\{[\alpha]_D +135.9^\circ$ (*c* 1.3, CHCl_3); ^1H -n.m.r. data (CDCl_3): δ 7.50–7.26 (m, 5 H, Ph), 5.36 (d, 1 H, $J_{1,2}$ 3.0 Hz, H-1), 5.14 and 5.04 (each d, 2 H, J_{gem} 1.0 Hz, OCH_2O), 4.78 and 4.74 (ABq, 2 H, J 12.0 Hz, CH_2Ph), 4.31 (d, 1 H, $J_{2,3}$ 10.0 Hz, H-3), 4.04 (q, 1 H, $J_{5,6}$ 6.0 Hz, H-5), 3.87 (dd, 1 H, H-2), 3.87 (s, 1 H, OH), 2.26 (s, 3 H, COCH_3), and 0.98 (d, 3 H, H-6)} via the corresponding diols* (**15**) in 55% overall yield, by successive reduction with lithium aluminum hydride and oxidation with *N*-chlorosuccinimide and dimethyl sulfide¹¹. Compound **16** was also obtained, in 50% yield, from **13** by treatment with mercuric oxide and mercuric chloride in aqueous methanol. Hydrogenation of **16** in the presence of palladium-on-charcoal gave the free sugar (**19**) as a syrup, $[\alpha]_D -45.2 \rightarrow -48.5^\circ$ (*c* 1.5, EtOH; 8 h), in 96% yield.

On the other hand, reaction of **11** with 1.5 molar equivalents of ethyltriphenylphosphonium bromide and butyllithium in ether gave a 1:1.6 mixture of the (*E*)- and (*Z*)-4-*C*-ethylidene derivatives (**21**) in 65% yield. Compound **21** was exclusively converted into the corresponding 4-*C*-acetyl derivative (**18**) having the D-*gluco* configuration¹² {m.p. $88\text{--}89^\circ$, $[\alpha]_D +152.7^\circ$ (*c* 2.6, CHCl_3); ^1H -n.m.r. data (CDCl_3): δ 7.50–7.30 (m, 5 H, Ph), 5.40 (d, 1 H, $J_{1,2}$ 3.0 Hz, H-1), 5.10 and 5.05 (each d, 2 H, J_{gem} 1.0 Hz, OCH_2O), 4.77 and 4.70 (ABq, 2 H, J 12.3 Hz, CH_2Ph), 4.24 (s, 1 H, OH), 4.10 (d, 1 H, $J_{2,3}$ 10.0 Hz, H-3), 3.95 (dd, 1 H, H-2), 3.79 (q, 1 H, $J_{5,6}$ 7.0 Hz, H-5), 2.26 (s, 3 H, COCH_3), and 1.07 (d, 3 H, H-6)} via the corresponding diols* (**17**), in 50% overall yield, by oxidation with 4-methylmorpholine *N*-oxide and a catalytic amount of osmium tetroxide, followed by oxidation as already described. Hydrogenation of **18** in the presence of palladium-on-charcoal gave the corresponding, free sugar (**20**) as a syrup in 96% yield, $[\alpha]_D -12.1 \rightarrow -15.3^\circ$ (*c* 1.7, EtOH; 8 h).

Thus obtained, **19** and **20** were oxidized with bromine in the presence of barium carbonate in water, and the resulting barium aldones were treated with Amberlite IR-120 ion-exchange resin in methanol, to give the corresponding methyl aldones; **2**, $[\alpha]_D -52.1^\circ$ (*c* 0.6, EtOH), and **4**, $[\alpha]_D -39.8^\circ$ (*c* 0.8, EtOH), as syrups, in 37 and 36% yield, respectively. It was found that the ^1H - and ^{13}C -n.m.r. parameters of **2** were completely identical with those reported² for methyl eurekaate, $[\alpha]_D -55.2^\circ$ (EtOH). Moreover, the ^1H -n.m.r. parameters of the monoacetate (**3**; m.p. $85\text{--}86^\circ$) of **2**, obtained

*For the examination of stereoselectivities in the reactions used, the (*S*)- and (*R*)-1-hydroxyethyl derivatives were separated. The results will be reported elsewhere.

by the usual acetylation of 2 with acetic anhydride in pyridine, were identical with those reported for methyl eurekanate monoacetate, m.p. 87°, and no depression of the melting point was observed on admixture of 2 with an authentic sample. From these results, the configuration of methyl eurekanate was determined to be *D-galacto***.

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REFERENCES

- 1 A. K. Ganguly, *Top. Antibiot. Chem.*, 2 (1978) 59–98.
- 2 W. D. Ollis, C. Smith and D. E. Wright, *Tetrahedron*, 35 (1979) 105–127.
- 3 A. K. Ganguly, O. Z. Sarre, A. T. McPhall, and R. W. Miller, *J. Chem. Soc. Chem. Commun.*, (1979) 22–24.
- 4 A. K. Ganguly, O. Z. Sarre, D. Greeves, and J. Morton, *J. Am. Chem. Soc.*, 97 (1975) 1982–1985.
- 5 M. Matsuzawa and J. Yoshimura, *Carbohydr. Res.*, 81 (1980) C5–C9.
- 6 J. S. Brimacombe, A. B. Foster, B. D. Jones, and J. J. Willard, *J. Chem. Soc., C*, (1967) 2404–2407; cf., K. S. Kim and W. A. Szarek, *Synthesis*, (1978) 48–49.
- 7 J. Yoshimura, K. Sato, and H. Hashimoto, *Chem. Lett.*, (1977) 1327–1330.
- 8 K. Sato, M. Matsuzawa, K. Ajisaka, and J. Yoshimura, *Bull. Chem. Soc. Jpn.*, 53 (1980) 189–191, and references cited therein.
- 9 H. Redlich, H.-J. Neumann, and P. Paulsen, *Chem. Ber.*, 110 (1977) 2911–2921; H. Paulsen and V. Sinnwell, *ibid.*, 111 (1978) 879–880.
- 10 A.-M. Sepulchre, B. Septe, G. Lukacs, and S. D. Gero, *Tetrahedron*, 30 (1974) 905–915.
- 11 E. J. Corey and C. U. Kim, *Tetrahedron Lett.*, (1974) 287–290.
- 12 D. L. Walker and B. Fraser-Reid, *J. Am. Chem. Soc.*, 97 (1975) 6251–6253.

**According to information from Prof. W. Keller-Schierlein of the Eidgenössische Technische Hochschule, Zurich, he determined the *D-galacto* configuration of methyl eurekanate monoacetate by X-ray analysis; paper submitted to *Helv. Chim. Acta*.